



Research Organization for Next Generation Vehicles

<https://www.waseda.jp/inst/nextgv/>

# Research on deNO<sub>x</sub> Catalysts for H<sub>2</sub> Internal Combustion Engines

Jin KUSAKA

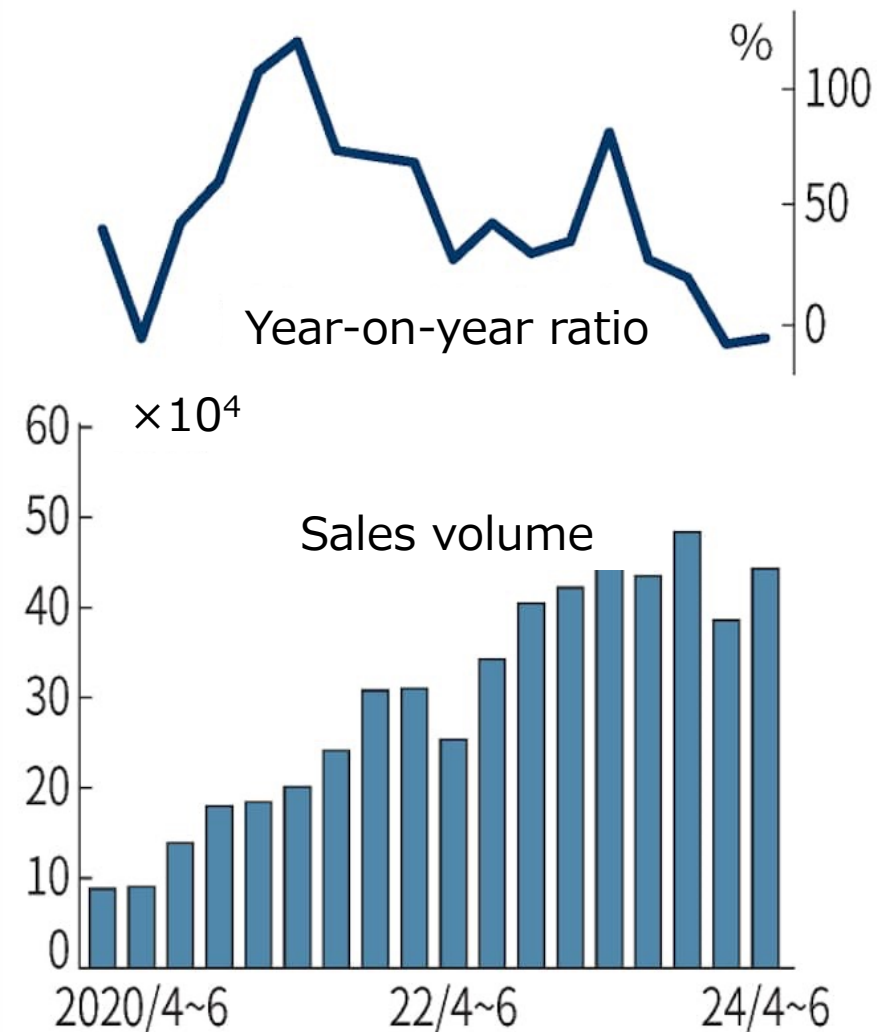
<https://jin.kusaka.w.waseda.jp>

Waseda University

# HEV is reversing to BEV in US

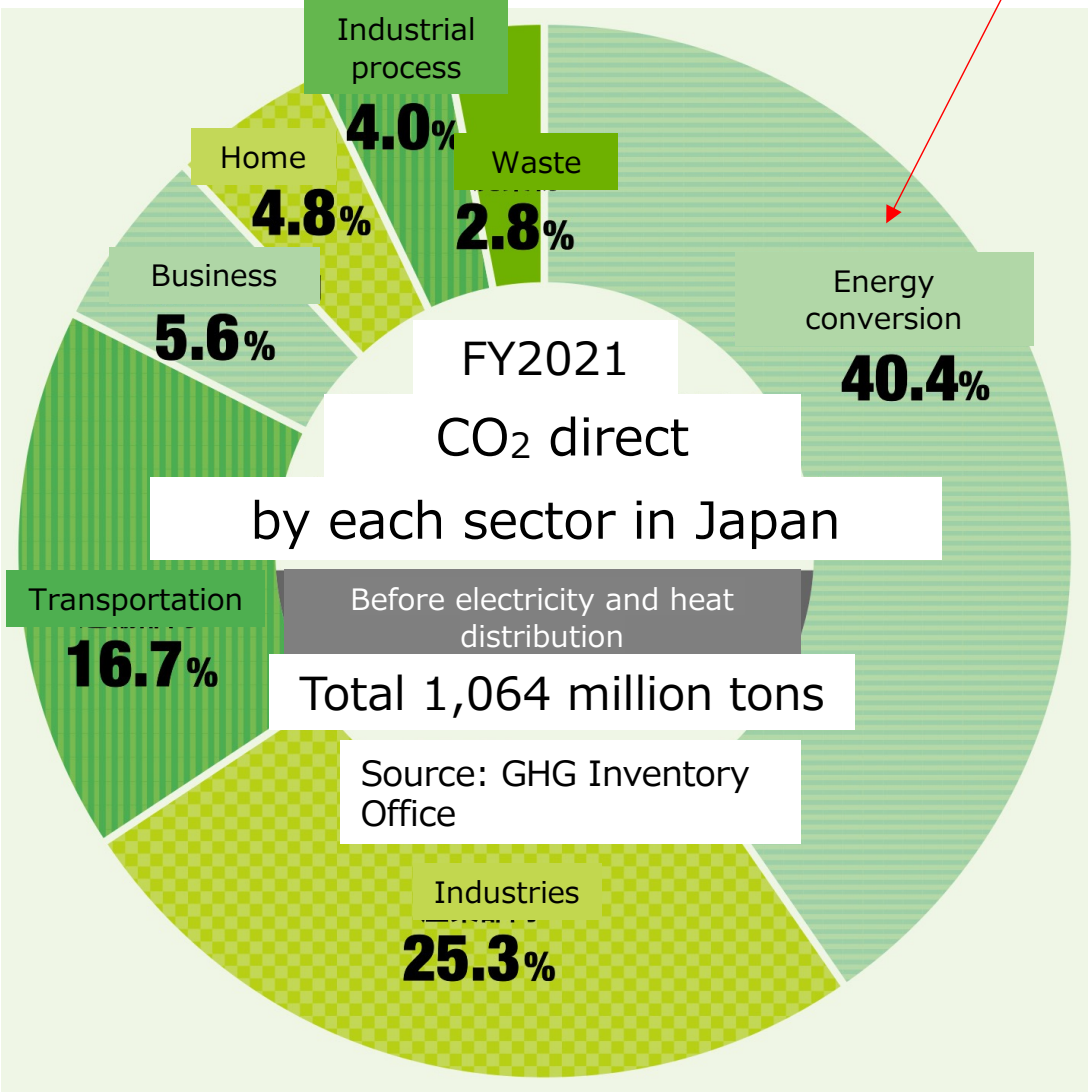
- BEV specialized model is at a crossroads
- World major market is HEV and PHEV
- GM is reviewing its EV strategy, which was based on the expansion of the EV market
- In-house development of Ford pickup EV truck once again is postponed
- The European Union (EU) initially had a policy of completely banning the sale of new engine vehicles in 2035, but sales continued as long as synthetic fuels were used

**Tesla global sales slow down clearly**

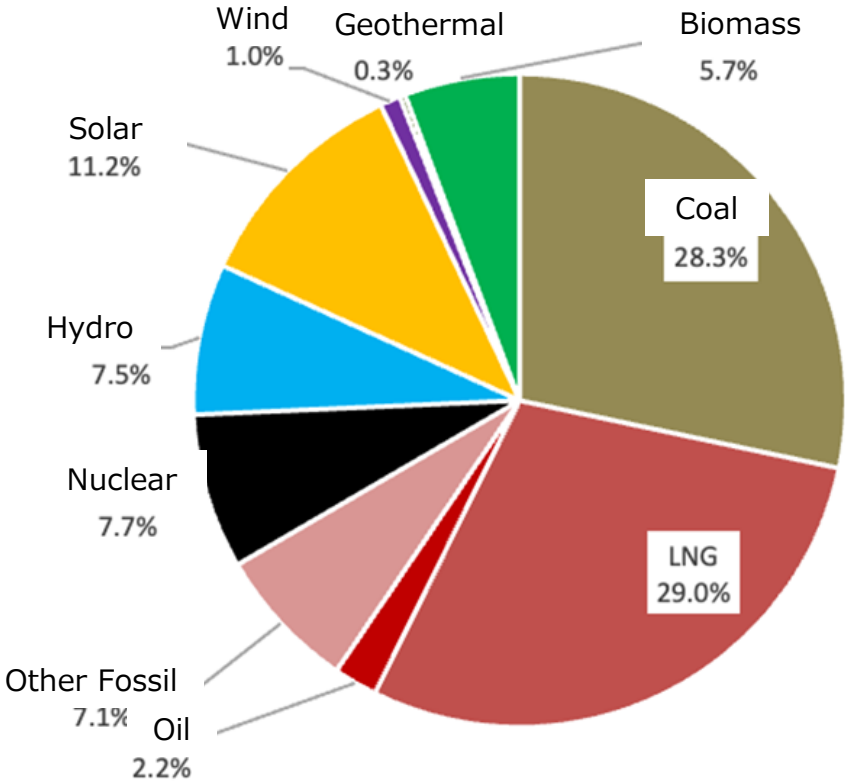


# CO<sub>2</sub> emissions in Japan in FY2021 (direct emissions)

In Japan, CO<sub>2</sub> emissions will not decrease no matter how many electric vehicles are used.

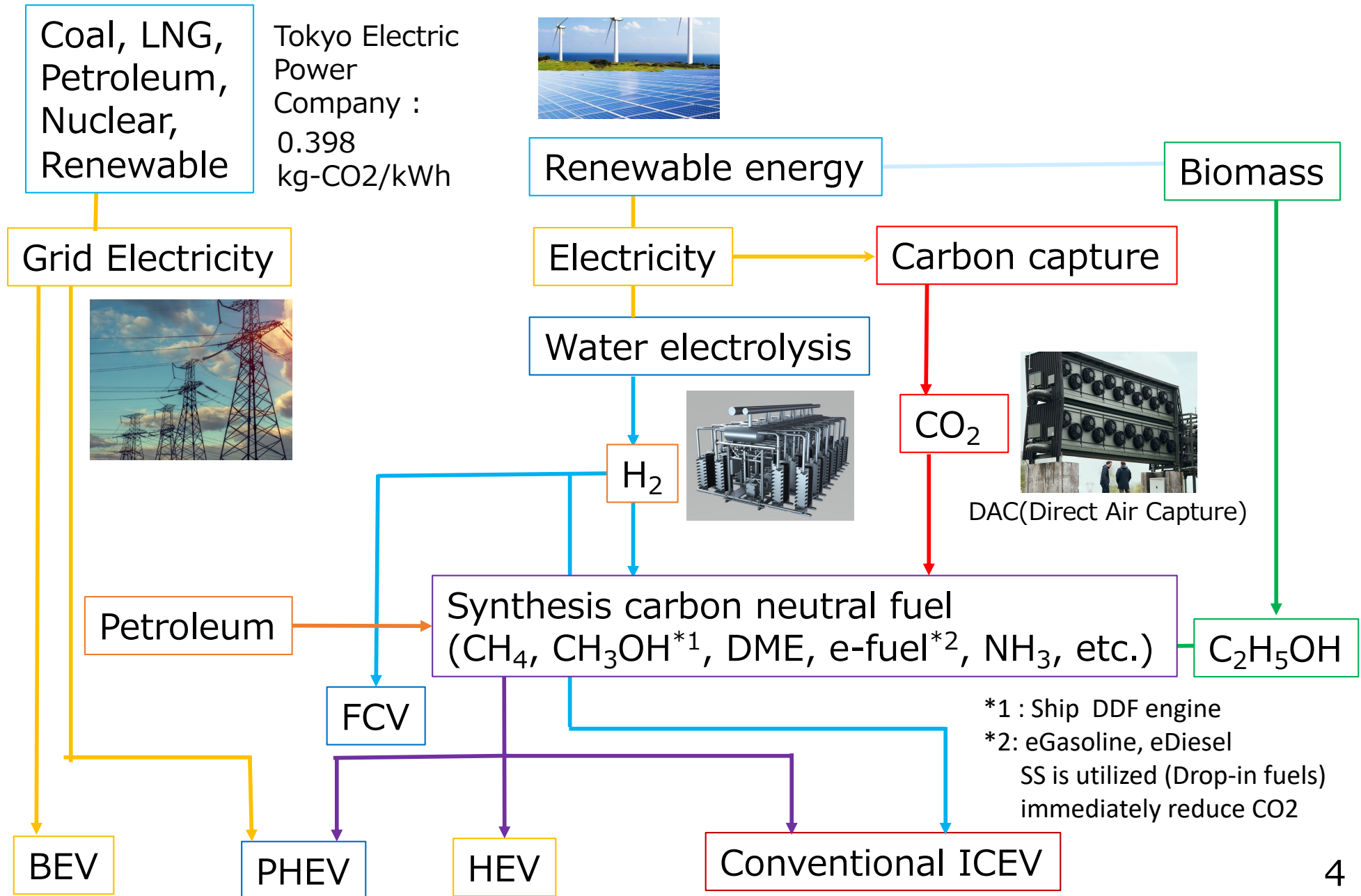


Renewable: 25.7%  
Nuclear: 7.7%



Source of power generation in Japan as preliminary data for 2023

# Carbon neutral energy and synthetic fuels for mobility

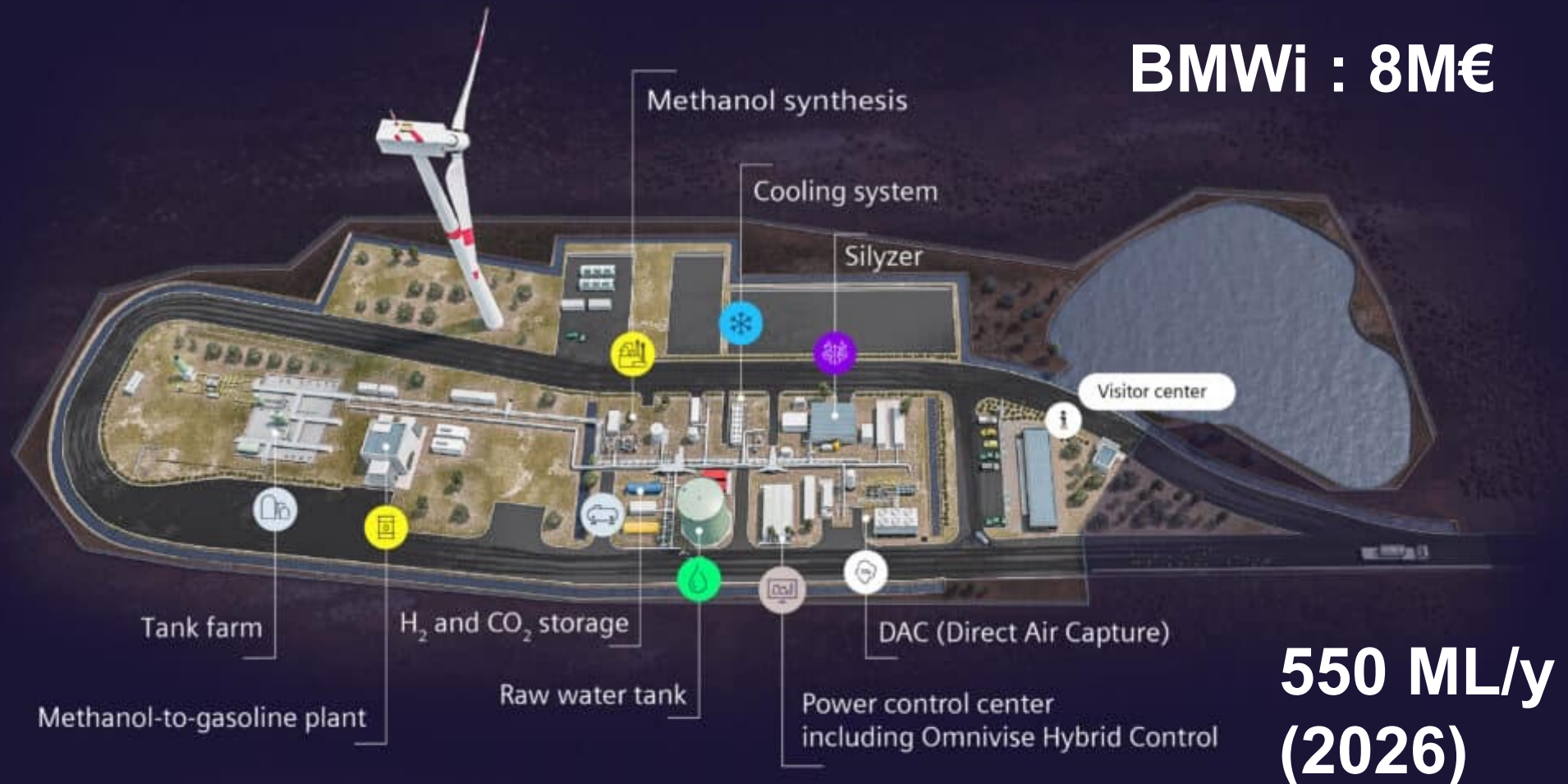


# Haru-Oni efuel(eGasoline) plant

5

Chilean power company AMF, HIF, Chilean national oil company ENAP, Italian Enel, German Siemens, **Porsche**, BMWi (German Ministry of Economy and Industry), US Exxon Mobile

<https://hydrogencouncil.com/ja/haru-oni-fuel-from-wind-and-water/>



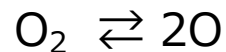
# MAN will introduce “hydrogen engine HD trucks in 2025

- Introduce heavy-duty truck hTGX with hydrogen combustion engine in 2025
- First 200 units manufactured
- Slight CO<sub>2</sub> emissions, but classified as zero-emission in Europe
- NO<sub>x</sub> increases in hydrogen combustion due to high temperature and lean mixture ( $2 < \lambda < 3$ )
- PN10 support
- Hydrogen infrastructure (number of SS and future pipeline) is also important

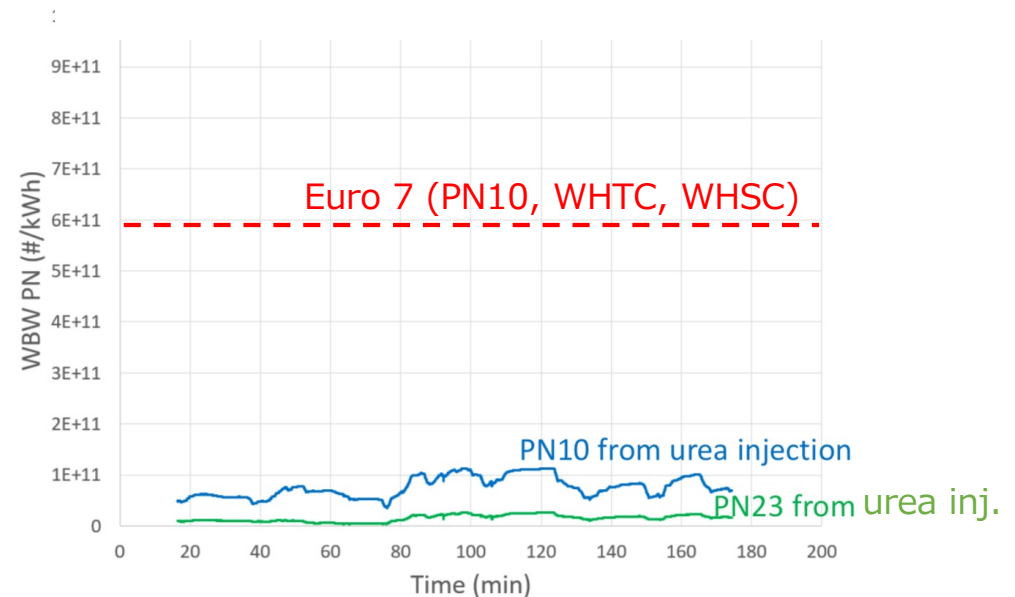
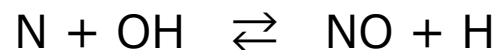
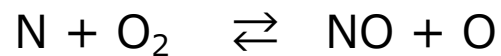
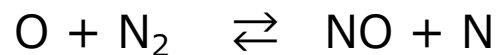


<https://bestcarweb.jp/fullload/839421>, 2024.4.13

Pyrolysis reactions at high temp.



Extended Zeldovich reactions



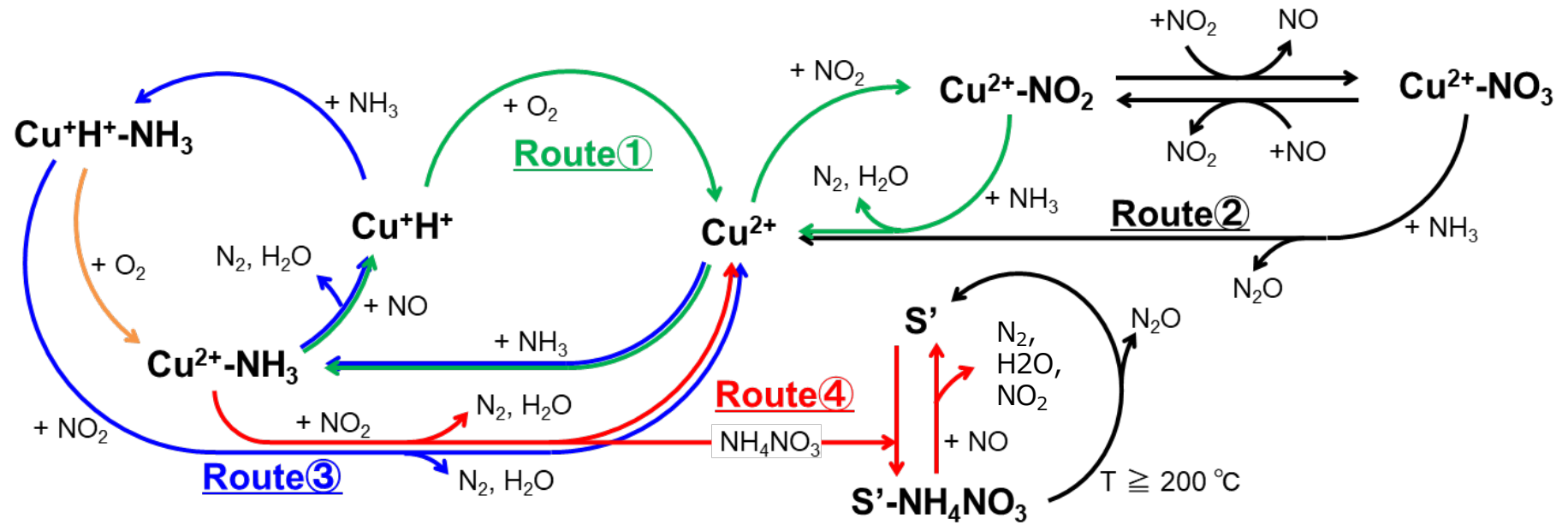
# Euro 7 and 6 limitations for NO<sub>x</sub>, PN<sub>10</sub> (PN<sub>23</sub>), PM, NH<sub>3</sub>, CH<sub>4</sub>, N<sub>2</sub>O, CH<sub>4</sub>, CO and NMOG for HDD<sup>[2]</sup>

Pollutant emissions	Euro 7 limits		Euro 6 limits	
	WHTC (CI/PI) and WHSC (CI) kWh	REAL Driving Emissions (RDE) kWh	WHTC (CI/PI) / WHSC(CI) kWh	PEMS (including. (CF)) kWh
<b>NO<sub>x</sub> mg</b>	200	260	460/400	690
<b>PN<sub>10</sub> Particles</b>	$6.00 \times 10^{11}$	$9.00 \times 10^{11}$	PN <sub>23</sub> $6.00 \times 10^{11}$ / $6.00 \times 10^{11}$	PN <sub>23</sub> $9.78 \times 10^{11}$
<b>PM mg</b>	8	-	10/10	-
<b>NH<sub>3</sub> mg</b>	60	(85)	10/10 (ppm)	-
<b>N<sub>2</sub>O* mg</b>	200	260	-	-
<b>CH<sub>4</sub>** mg</b>	500	650	500 (PI only)	750 (PI only)
<b>CO mg</b>	$1.50 \times 10^3$	$1.95 \times 10^3$	$4.00 \times 10^3$ / $1.50 \times 10^3$	$6.00 \times 10^3$
<b>NMOG mg</b>	80	105	160 / 130 (CI: THC, PI: NMHC)	240 (CI: THC, PI: NMHC)

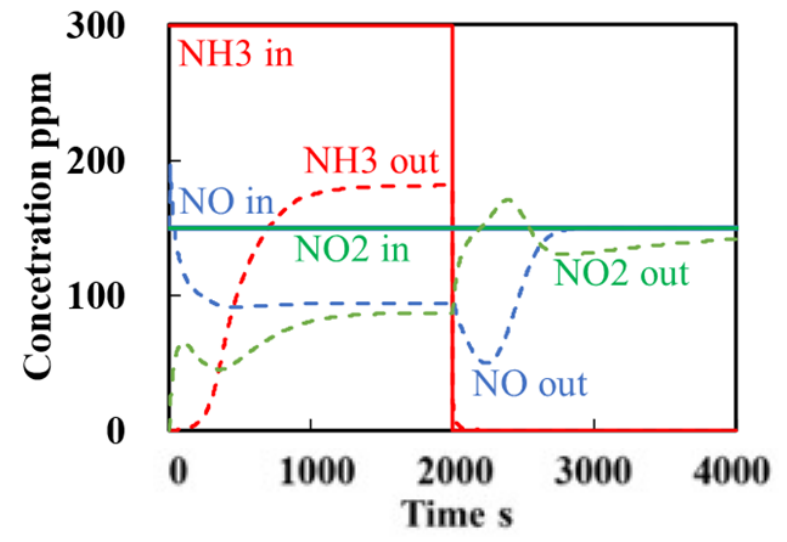
\* Global warming potential : 310

\*\* Global warming potential : 25

# Reaction mechanism of Cu-Chabazite SCR [1], [3], [4]

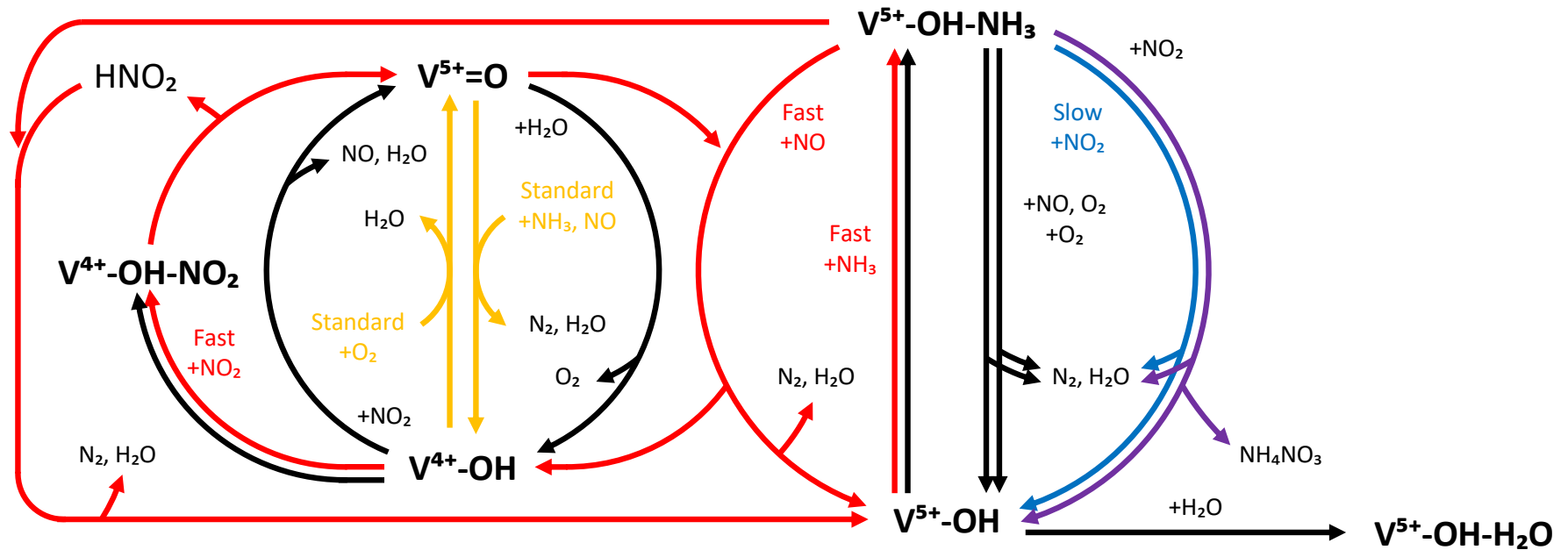


- High  $\text{NO}_x$  conversion rate
- Much  $\text{NH}_3$  storage
- $\text{NH}_3$  slip
- Formation for  $\text{S}'\text{-NH}_4\text{NO}_3$
- $\text{N}_2\text{O}$  desorption





# Reaction mechanism of vanadium SCR<sup>[5]</sup>



[6] Xiangmin Wang, Xuesen Du, Guangpeng Yang, Jingyu Xue, Yanrong Chen and Li Zhang, Chemisorption of NO<sub>2</sub> on V-Based SCR Catalysts: A Fundamental Study toward the Mechanism of "Fast SCR" Reaction, 2019

[7] Topsoe, N. Y. Mechanism of the selective catalytic reduction of nitric oxide by ammonia elucidated by in situ on-line fourier transform infrared spectroscopy. Science 1994,265, 1217–1219

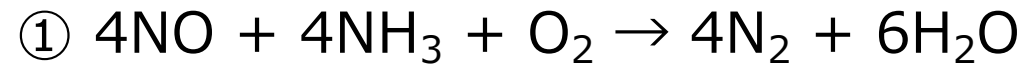
[8] Nova, Ciardelli, Tronconi, Chatterjee, BandlKonrad, NH<sub>3</sub>-NO/NO<sub>2</sub> chemistry over V-based catalysts and its role in the mechanism of the Fast SCR reaction, 2006, 112, 3-12

# Cu-zeolite vs Vanadium

	Cu-zeolite	Vanadium
Conversion(purification) rate	○	
NH <sub>3</sub> storage	○	
N <sub>2</sub> O formation		○
Sulfur poisoning		○
Thermal resistance	○	
Price		○

# Carry over Diesel EATS for EU7

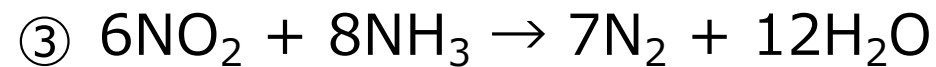
## Urea SCR catalyst



Standard SCR reaction

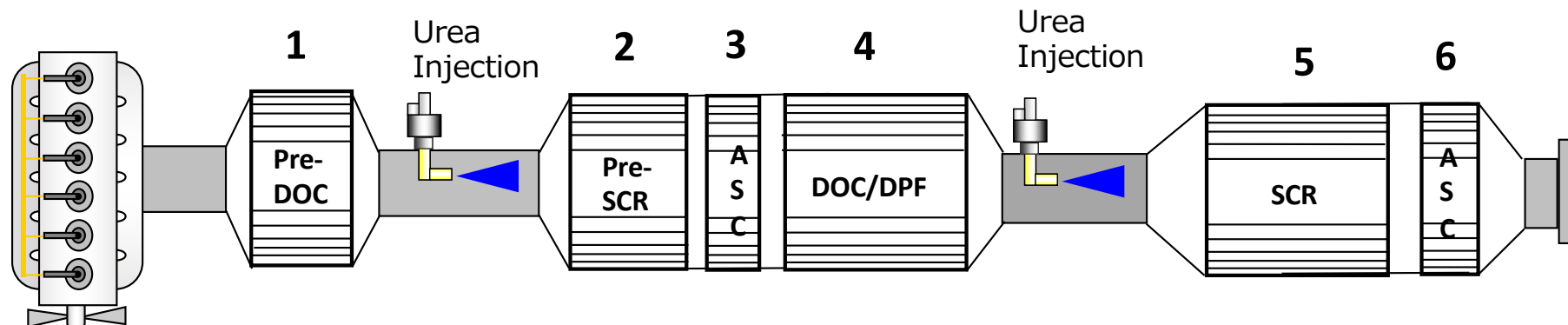
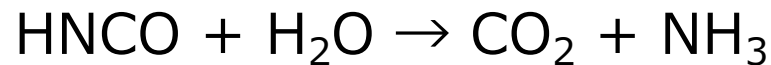
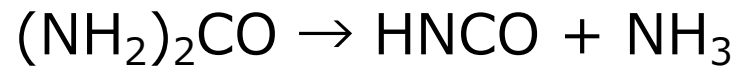


Fast SCR reaction

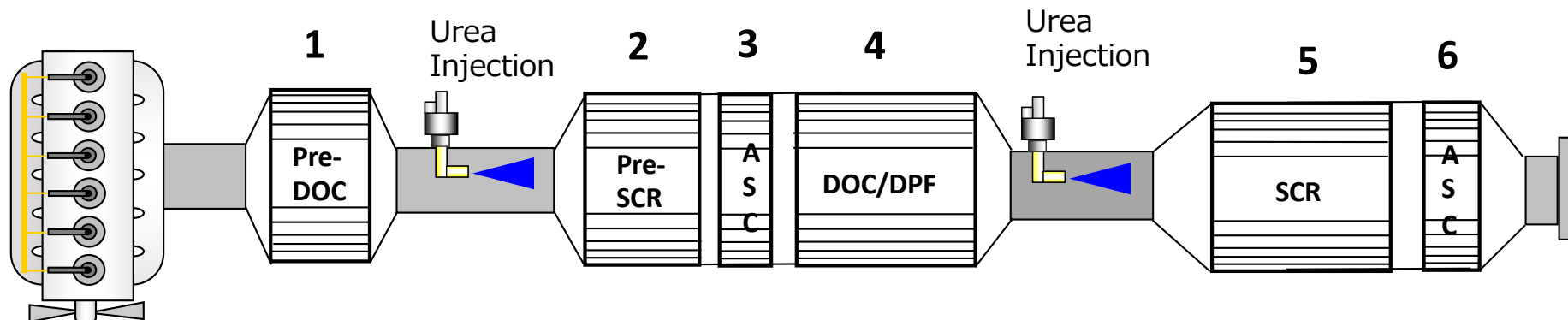


$\text{NO}_2$  SCR reaction

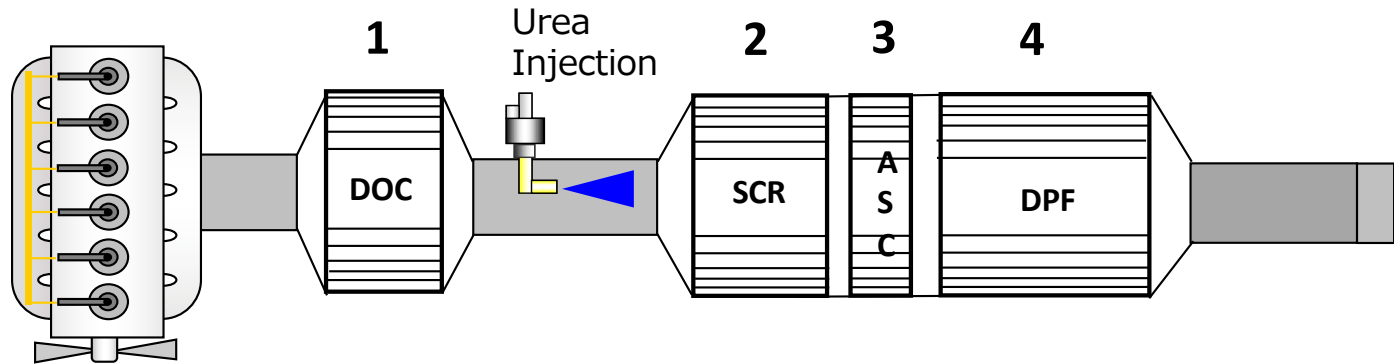
## Urea Pyrolysis, hydrolysis reactions



- 1 Pre-DOC : Exothermic reaction by  $H_2$  supply  
 $NO_2/NO_x$  ratio(NO oxidation)
- 2 Pre-SCR + ASC:  $NO_x$  conversion and  $NH_3$  slip removal  
ASC: Ammonia Slip Catalyst
- 3 DOC/DPF: Filtration for PN (urea born, lubricant)  
NO oxidation
- 4 SCR :  $NO_x$  conversion
- 5 ASC : Ammonia removal



# Dedicated H<sub>2</sub> engine SCR



1 DOC : NO<sub>2</sub>/NO<sub>x</sub> ratio (NO oxidation)

2 SCR: Cu-zeolite or Vanadium

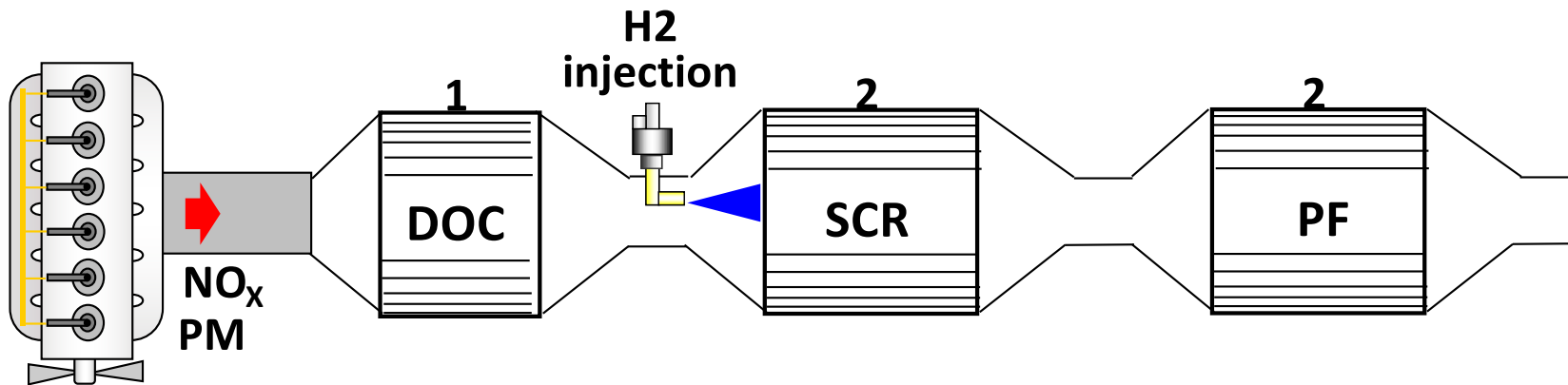
3 ASC: Ammonia removal

4 DPF: PN10(Urea born, lubricant) removal

# SCR catalyst using hydrogen as reductant

It is preferable not to have a urea water tank.

If zeolite or vanadium is used, nonpolar  $H_2$  is most likely not adsorbed. Therefore, We will explore new active metals. The  $H_2$  injection control is expected to be more important.



SCR: Catalytic active metals are newly explored

# H<sub>2</sub> engine exhaust (WHTC)

- High H<sub>2</sub>O concentration ( ~30%)
- Exhaust temp. (room temp. ~400 deg.C)
- Excess air ratio ( $1.5 < \lambda < 3.0$ )
- NO/NO<sub>2</sub> ratio (2 ~40)

# Key points of R&D

- Ensuring purification performance of Cu zeolite and vanadium catalysts under conditions of H<sub>2</sub>O mole fraction up to 30% and high temperature
- Development for dosing strategies of reducing agent
- Unravel the hydrothermal degradation mechanism under high-temperature exhaust gas equivalent to that of a spark ignition engine and its countermeasures
- Development of a new catalyst that uses H<sub>2</sub> as a reducing agent
- Construction of a numerical calculation model



# Next-generation heavy duty vehicle development promotion project by industry-academia collaboration for decarbonization (6th phase)

脱炭素に向けた産学官連携による次世代大型車両開発促進事業（第6期）



- A mini-reactor experiments
- Modeling
- Dosing strategies
- Performance and mechanism feedback

Waseda University

Catalyst company

- Reactor experiments
- Durability analysis
- Test piece
- Full-size catalyst

Automobile company

- Actual engine test
- Catalyst evaluation

# Thank you for your attention

## Reference

1. Fuka Yoshida et al., Acceleration of Fast-SCR Reaction by Eliminating The Ammonia Blocking Effect, **SAE 2024-37-0001**, TORINO, CO<sub>2</sub> Reduction for Transportation Systems Conference, 12-13<sup>th</sup>, June, 2024
2. Jin Kusaka, Fuka Yoshida, MILT Industry-Academia Collaboration Project (5th Phase) Results , JSAE Forum, 22<sup>th</sup>, July, 2024
3. Fuka Yoshida, Jin Kusaka, et al, Urea Pulse Injection Strategies for Improvement of NOx Reduction in the SCR System, JSME Annual Congress(Ehime), 11<sup>th</sup>, September, 2024
4. Fuka Yoshida, Jin Kusaka, et al, Experimental and numerical investigations on the effect of urea pulse injection strategies to reduce NOx emission in urea-SCR catalysts, **SAE 2024-01-4304**, 13<sup>th</sup>, November, 2024
5. Yuya Kotani, Mai Shinkawa, Research progress report for the 4<sup>th</sup> Research and Development Meeting, 2024, August

## Contact

Jin Kusaka, Dr. Eng., Professor

